ORGANIC CHEMISTRY



DPP No. 3

Total Marks: 32

Max. Time: 35 min.

Topic: IUPAC Nomenclature & Isomerism

Type of Questions

Single choice Objective ('-1' negative marking) Q.1 to Q.4

Multiple choice objective ('-1' negative marking) Q.5 to Q.6

Subjective Questions ('-1' negative marking) Q.7

Match the Following (no negative marking) Q. 8

1. Which of the following alkanes contains primary, secondary, tertiary and quaternary carbon atoms together.

(A) (CH₂)₂CH

(B) $(C_2H_5)_2CH$

(C) (CH₂)₂CCH₂CH(CH₂)₂

(D) (CH₂)₄C

M.M., Min. [12, 12]

[8, 8]

[4, 5]

[8, 10]

2. Which of the following has longest chain of carbon:

$$\begin{array}{c} \operatorname{CH_2}-\operatorname{CH_3} \\ \text{(A) } \operatorname{CH_3}-\operatorname{CH}-\operatorname{CH_2}-\operatorname{CH}-\operatorname{CH_3} \\ \text{CH}_2-\operatorname{CH_3} \end{array}$$

$$\begin{array}{c} \text{CH}_{3} \\ \text{(D)} \ \text{CH}_{3} - \text{CH} - \text{CH}_{2} - \text{CH}_{2} - \text{CH} - \text{CH}_{2} - \text{CH}_{3} \\ \\ \text{CH}_{2} - \text{CH}_{3} \end{array}$$

(3 marks, 3 min.)

(4 marks, 4 min.)

(4 marks, 5 min.)

(8 marks, 10 min.)

3. In following compound -

The correct lowest set of locants are

(A) 3,3,4,5

(B) 3,4,5,5

(C)4,5,3,3

(D) 5,5,4,3

4. The correct IUPAC name of the following compound is

$$\begin{array}{c|c} \operatorname{CH}_3 - \operatorname{CH} - \operatorname{CH} - \operatorname{CH}_2 - \operatorname{CH}_3 \\ | & | \\ \operatorname{CH}_2 & \operatorname{CH} - \operatorname{CH}_3 \\ | & | \\ \operatorname{CH}_3 & \operatorname{CH}_3 \end{array}$$

(A) 4–Ethyl–3,5–dimethylhexane

(C) 3-Isopropyl-4-methyhexane

(B) 2,4-Dimethyl-3-ethylhexane

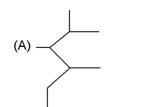
(D) 3-Ethyl-2,4-dimethylhexane

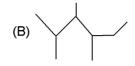


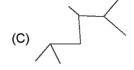




5.* The correct structure of 2,3,4-Trimethylhexane is:







6.* Choose the correct option's according to given IUPAC name:

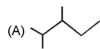
- (A) Neohexane is 2,2-Dimethylbutane.
- (B) Isobutane is 2-Methylpropane.
- (C) Isopentane is 2-Methylbutane.
- (D) Neopentane is Dimethylpropane.

7. Calculate the molecular weight of the lowest hydrcarbon which contains sp & sp2 hybridised carbon atoms only.

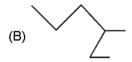
8. Match the following:

Column-I

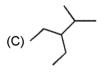
Column-II



(p) 3-Ethyl-2-methylpentane



(q) 3-Methylhexane



(r) 2,3-Dimethylbutane



(s) 2,3-Dimethylpentane

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(A)

- 1. (C)
- 3.
- (D)

- 5.* (A,B,D)
- (A,B,C,D)

(C)

 $H_2C = C = CH_2$ 7. M. W. = 40.

8. [A-s]; [B-q]; [C-p]; [D-r]

Hints & Solutions

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CH₃ CH₃
3. (A)
1
CH₃ $^{-2}$ CH₂CH₂CH₂CH₋CH₋CH₋CH₂CH₃

CH₃ CH₃

Lowest set of Locant (3, 3, 4, 5)

